



Review

Therapeutic Perspectives of Metal Nanoformulations

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Abstract: In recent decades, acceptance of nanoparticles (NPs) in therapeutic applications has increased because of their outstanding physicochemical features. By overcoming the drawbacks of conventional therapy, the utilization of metal NPs, metal-oxide, or metal supported nanomaterials have shown to have significant therapeutic applications in medicine. This is proved by a lot of clinical and laboratory investigations that show improved treatment outcomes, site-specific drug delivery, and fewer side effects compared to traditional medicine. The metal NPs interaction with living cells (animal and plant) showed many ways to develop therapeutic models with the NPs. Despite all of the advancements that science has achieved, there is still a need to find out their performance for long-term use to solve modern challenges. In this regard, the present documentation reviews some potential metals, including silver (Ag), gold (Au), zinc (Zn), copper (Cu), iron (Fe), and nickel (Ni) NPs, as therapeutic agents in various areas such as anticancer, antimicrobial, antidiabetic, and applicable for the treatment of many other diseases. Depending on the outstanding ongoing research and practical trials, metal-based NPs can be considered the hope of prospective modern therapeutic areas.

Keywords: metal nanoparticles; nanotherapy; therapeutic uses; targeted strategy



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1. Introduction

The implementation of nanoparticles (NPs) for the treatment and diagnosis of disease is a revolutionary concept that has been developed over the past few decades. The nanotechnological approach can be divided into two branches: one is nanodevices and the other is nanomaterials. The nanodevice can be defined as such tiny devices at the nanoscale range, which includes microarrays and some devices such as respirocytes [1–3].

Particles smaller than 100 nanometers (nm) in any one of the dimensions are considered nanomaterials. Biomedical science found successful result by using nanoparticles as therapeutic agents in the treatment of various diseases. As it is selective on the target organ and receptors, it overcomes several limitations of conventional therapy, such as nonspecificity, unwanted side effects, less efficiency, and low bioavailability [4].

Therefore, current research projects are considerably more focused on developing and designing new drug delivery systems, and the most promising area is ensured by NPs for their uniqueness in biological and physicochemical characteristics, as they can deliver molecules to specific locations in the body [5].

The therapeutic molecules which are insoluble in water can be complexed with NPs, resulting in greater bioavailability and significantly fewer physiological barriers; for example, NP carriers assist medication in passing the blood–brain barrier (BBB) [6–8]. Nevertheless, because it is targeted, it will require lower doses than conventional therapy, and the therapeutic index will be higher as it will minimize the toxicity in the biological system. The utilization of NPs in various fields of the health sector is possible because of their ability to provide a visual image of the targeted delivery location by using some agents; moreover, their pathway can be tracked.

Several studies have recently focused on the method of producing metal NPs using green synthesis, which has shown positive results against pathogens, cancer cells, helminths, fungi, etc. using the metal NPs Zn, Ag, Au, Pt, Mn, Ni, and Ti [9]. Currently, among other NPs, Ag-NPs are one of the top listed compounds being researched [10]. In 1857, Michael Faraday was the first person to study Au-NPs in a colloidal system and report Au-NP's optical features [11].

This documentation reviewed physicochemical properties of the NPs Au, Ag, Cu, Fe, Ni, and Zn and generated a review of the recent year's progression for their use as nanomedicine with their application as targeted delivery in numerous physical disorders.

1.1. Significance of Metals in Human Body

For diagnostic and treatment purposes, metals are used in organic systems as medicine [12]. There are some metals, such as mercury (Hg), cadmium (Cd), and arsenic (As), that are considered toxic when they cross a certain limit, but some metals are necessary for the body's enzymatic and metabolic functions. For instance, there are metalloproteins which contain metal ions as co-factors, and a greater portion of proteins are in this category. There is a minimum 1000–3000 human proteins which contain a Zn ion as a co-factor [13]. The comparative presence of metals in the human body shown in Table 1.

Table 1. Concentration of metals in some human organs [14].

Metals	Liver (ppm)	Kidney (ppm)	Lung (ppm)	Heart (ppm)	Brain (ppm)	Muscle (ppm)
Iron	16,769	7168	24,967	5530	4100	3500
Manganese	138	79	29	27	22	<4–40
Nickel	<5	<5–12	<5	<5	<5	<15
Zinc	5543	5018	1470	2772	915	4688
Cobalt	<2–13	<2	<2–8	-	<2	150
Copper	882	379	220	350	401	85–305

1.2. Major Function of Metals in Human Body

Sections 1.2.1–1.2.7 aim to describe the significance of the selected metals in the general physiological functions of the human body.

1.2.1. Manganese (Mn)

Manganese is significant for development, metabolism, and the antioxidant system. Importantly, Mn is needed for amino acid, cholesterol, and carbohydrate metabolism and in bone and thyroxin formation.

Mn is required for the action of enzyme families, such as oxidoreductases, hydrolases, transferases, lyases, isomerases, and ligases. For normal immune protection, blood sugar control, creating cellular energy, and reproduction, Mn works with various organ systems. It is estimated by the National Research Council that for adults, 2 to 5 mg of dietary manganese per day is safe [15].

It is particularly important for the detoxification of superoxide free radicals, and it activates some metalloproteases. It assists the body in using biotin, thiamin, vitamin C, and choline. However, excessive intake can lead to a stage called manganism, which causes neuronal death and a Parkinson's-like syndrome [16].

1.2.2. Iron (Fe)

Fe is a vital component of hemoglobin (RBC) [17]. Fe aids in the metabolism of muscle and active connective tissue. It is needed for the synthesis of some hormones, neurological development, maintaining physical growth, and cellular functioning [18,19].

For the synthesis of DNA and electron transportation, Fe is important [20]. Fe deficiency is the reason for about 50% of the cases of anemia around the world, according to a WHO report [21].

1.2.3. Cobalt (Co)

Co is a part of cobalamin, or vitamin B12, and therefore, it is significant for the function of cells. Co is needed for the production of RBC and the production of antimicrobial compounds (antibacterial and antiviral). Cobalt plays a vital role in amino acid and protein generation and the formation of neurotransmitters. Co salt is used in the treatment of anemia [22].

1.2.4. Nickel (Ni)

For regulating the proper function of the human body, Ni is an essential micronutrient. This metal amplifies hormonal function and is also required in lipid metabolism [23]. Although the mechanism of toxicity is unknown, prolonged contact or higher intake can result in a variety of side effects, including cancer [24]. Ni is required in trace amounts for growth and reproduction [25]. It activates arginase and urease enzymes [12] and also inhibits some enzymes, for instance, acid phosphatase [26].

1.2.5. Copper (Cu)

Cu is highly involved in energy production, iron metabolism, the formation of connective tissue, and the activation of neuropeptides and neurotransmitters [17,18]. Ceruloplasmin (CP), a Cu-abundant enzyme involved in Fe metabolism, is mostly composed of Cu and accounts for approximately 95% of total Cu in human plasma [27]. Cu is also involved in various physiologic processes, including angiogenesis, brain development, pigmentation, neurohormone homeostasis, gene expression regulation, and immune system functioning [24], as well as providing protection against oxidative damage [28,29].

1.2.6. Zinc (Zn)

Zn is required in a variety of ways for cellular metabolism. Zn is mandatory for the activation of about 100 enzymes [30,31], and it has roles in the immune system, protein synthesis [32,33], wound healing [34], DNA synthesis, and cell division [35,36]. It assists the normal development of the fetus during pregnancy and is needed for further growth from the stage of childhood to adolescence [37,38]. Moreover, Prasad et al. demonstrated that Zn is responsible for a proper sense of smell and taste [39,40]. Because the human body cannot store zinc, it must be consumed on a regular basis to keep these functions running smoothly [41].

1.2.7. Gold (Au)

The average human body (for an average adult human weighing 70 kg) might contain about 0.2 mg of Au [42]. Significant health functions include helping to maintain our joints as well as facilitating the transmission of electrical signals throughout the body. It is necessary for the maintenance and function of the joints. Additionally, Au is an excellent conductor of electricity, aiding in the transmission of electrical signals throughout the body [39]. Several cell-mediated immune responses to various mitogens and antigens are inhibited by gold compounds. This inhibition is accelerated by the Au's impact on macrophages [43].

1.3. Size, Shape, Material, and Surface of Nanoparticles

NPs range from 1 to 100 nm (Figure 1) and could be a sphere, cube, rod, plate, or star shape.

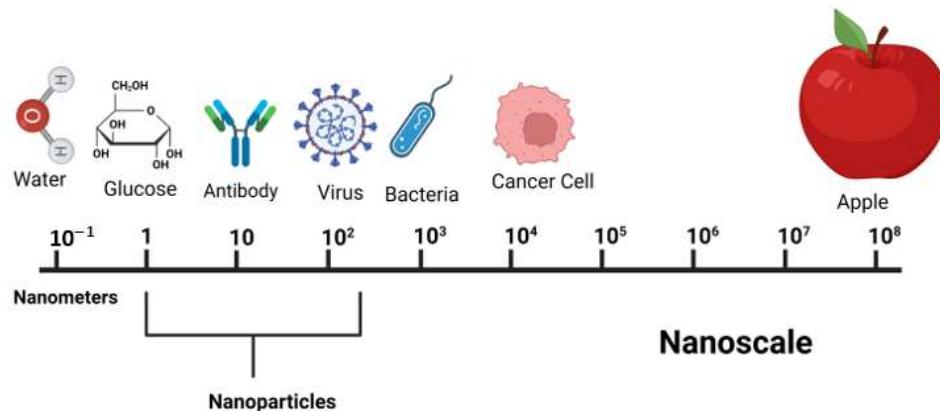


Figure 1. Sizes of nanoscale items in comparison to other relevant objects.

The surface of the NPs can be PEGylation or another coating, which might have present linkers containing surface functional group, surface charge, and targeting ligand (antibody, peptide, aptamer, etc.); see Figure 2. Nanomaterial size, shape, and surface coating are essential parameters that influence cell uptake and/or the pace and site-specific drug delivery from the system. The shapes of nanoparticles also play a crucial role in infrared absorption, which is particularly essential in phototherapy [44]. Rods are the most absorbent, followed by spheres, cylinders, and cubes [45].

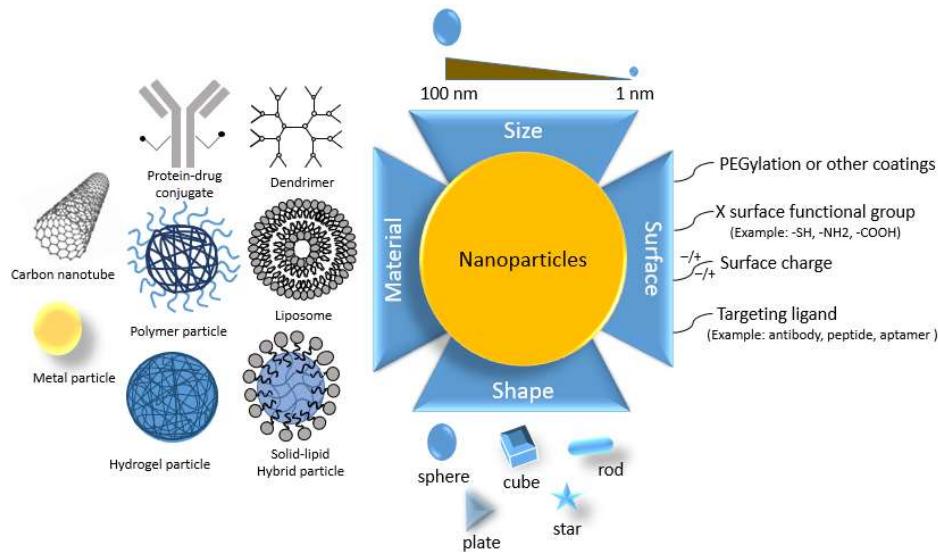


Figure 2. Size, shape, and surface material of nanoparticles.

1.4. Major Nanodrug Delivery Systems

This section will familiarize you with the various types of nanomedicine and provide a general idea for further research. Based on the recent approaches, polymeric, metallic, and ceramic NP drug delivery vehicles are widely used [8], such as liposomes [46], micelles [47], dendrimers [48], etc. A large number of clinical and pre-clinical trials demonstrated their efficacy in treating various diseases [49–51].

A process through which cells take in foreign material by enveloping it with their membrane is known as endocytosis. Pinocytosis and phagocytosis are the two main subtypes of endocytosis. Hormonal receptors, integrins, growth factor receptors, tyrosine

kinase receptors, and lipids are just a few of the proteins that are transported via the critical cellular process known as endosomal trafficking.

Pinocytosis, from the Greek “*pino*” meaning to drink, is the mechanism through which the cell absorbs liquids and disperses tiny molecules. The cell membrane bends and forms tiny pockets during this process, catching the cellular fluid and other dissolved materials (Figure 3). In many cases, nanodrugs follow this endocytosis (pinocytosis) [52]. Other delivery methods include clathrin-dependent delivery [53,54].

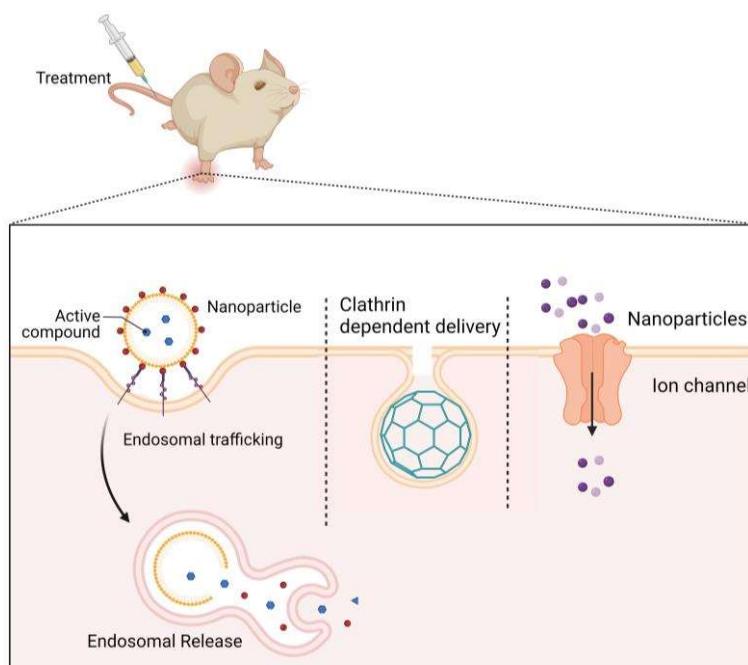


Figure 3. Major cellular uptake methods of nanoparticles (cellular uptake occurs mainly through endosomal trafficking, through clathrin-dependent delivery, and through ion channels).

Biomedical uses of nanohydrogels are wide in drug administration, tissue engineering [55], and wound dressing and healing due to their biocompatibility [56], nontoxicity [57], and high absorption capacity [58]. Furthermore, site-specific targeted drug administration is possible with stimulus response factors such as temperature and pH-dependent upgraded hydrogels [52,59]. There is evidence of the use of nanometal-hydrogel for tissue regeneration [60].

Nanohydrogel molecules have the features of hydrophilic and hydrophobic components, disperse in the solution to form micelles [61]. Micelles are generated by self-assembly, where the process does not begin until a specific minimum concentration is reached. This concentration is frequently referred to as the crucial micellar concentration [62]. Ag-NPs form micelles to be stable in aqueous solutions [59]. Reverse micelles are used for bimetallic (Au/Pd) NP formation [63].

Dendrimers are such structures that have branches or arms like trees and are globular, nanodimensionally compact, and radially symmetric [64]. The capacity of dendrimers to distribute drugs in a regulated and targeted manner is their most promising use. Higher stability, a longer half-life, and greater bioavailability are characteristics of drugs conjugated to such delivery systems. Additionally, prolonged drug release via the drug-dendrimer combination lowers the systemic toxicity and maintains tumor tissue-specific aggregation [65,66].

A significant number of clinical and preclinical studies show how deeply the function of NPs as carriers of therapeutic agents has been studied. NPs are regarded as one of the most promising groups of medication delivery systems. NPs can bind macromolecules such as proteins, antibodies, or nucleic acids and can encapsulate both hydrophilic and

hydrophobic medicines [67]. Paclitaxel has been exemplified in polymeric NPs made by impeding copolymers of mono-methoxy polyethylene glycol and poly-D,L-lactide [68].

Additionally, NPs may be programmed to react to many environmental factors, including pH, light, temperature, enzymes, and other biological and chemical agents. The most often employed of all these stimuli is pH responsiveness. The pH differential can aid in distinguishing tumor tissue (pH 5.7–7.0) from normal tissue (pH 7.4). The capacity to directly release medications at tumor sites has made this pH responsiveness valuable in a variety of cancer and tumor therapies [69].

Inorganic NPs have been studied for their potential biomedical applications in addition to polymeric NPs. There are various ways to make inorganic NPs, including the crystallization of inorganic salts, thermal breakdown, and other well-known synthetic processes [70]. Several inorganic NPs, including Au, Ag, Pt, iron oxide (FeO), cerium oxide (CeO_2), and zinc oxide (ZnO), have been successfully synthesized and used in numerous preclinical and clinical trials. However, because of their higher biocompatibility, higher biodegradability, and lower systemic toxicity, polymeric nanoparticles are preferred over inorganic NPs [71]. Many researchers have been interested in liposomes as a potential medication delivery technology due to their capacity to selectively transport both hydrophilic and hydrophobic medicines to their respective target sites [72].

Liposomes are capable of encapsulating and protecting both hydrophilic and hydrophobic medicines before releasing them at specific sites. Multilamellar vesicles are made up of concentric spheres of phospholipids separated by water layers, while unilamellar vesicles only have a single phospholipid bilayer encapsulating the aqueous solution [50]. Au-NPs enable green synthesis by using glycerol liposomes [73,74] and the selective release of contents from liposomes caused by light [75]. Multifunctional metallic NPs can be formed for medical imaging and micro-fluidity [76,77].

Scaffolds are important in biomedicine and tissue engineering because of their capability to foster cell adhesion, proliferation, and differentiation, all of which are necessary for tissue development. The scaffolding allows cells to develop in all the right places, which results in the production of tissue. A biocompatible matrix is required for optimal cell growth, the strategy of employing scaffolds is of utmost relevance [78,79]. Scaffolds have great biocompatibility, mechanical strength, porosity, and interconnectivity, all of which are necessary for clinical application [80]. There is evidence of a nanohybrid scaffold of glycolic acid-g-chitosan-Pt- Fe_3O_4 being used as a drug delivery system [81].

1.5. Scopes of Metal Nanoparticles in Remedies

To ensure that the human body functions normally, specific levels of certain metals must be present. The main functions of metals are to catalyze certain reactions and act as cofactors or prosthetic groups of enzymes. The required metals for humans include Na, K, Fe, Mg, Zn, Cd, Mn, Cu, V, Cr, Mo, Co, and Ni. In the absence of certain essential metals, anemia could occur [41].

Iron deficiency causes the loss of functional blood proteins such as hemoglobin, myoglobin, etc., whose function is to carry oxygen. Iron deficiency accounts for roughly half of all anemia cases worldwide. As a first-line therapy, oral iron supplementation is recommended; however, IV iron formulation is a recent addition to anemia treatment, and hepcidin could be a future diagnostic target [42]. Vitamin B12 is made of a cobalt complex called cobalamin, and the lack of this vitamin results in pernicious anemia.

Zn is used as a catalyst for various enzymes. Importantly, it is required for red blood cell production. That is why a deficiency in this metal can cause anemia. It can heal wounds, and Zn ions (Zn^{2+}) can be used for treating the herpes virus [82]. According to one study, infant's diets which had low in Zn; had higher rates of copper anemia, which can lead to heart disease also [31]. Copper gluconate, copper chloride, or copper sulfate are used as oral or IV copper supplements in copper anemia [83]. Some potential metallic NPs are shown in Figure 4.

For the treatment of rheumatoid arthritis [84], juvenile rheumatoid arthritis, and psoriatic arthritis, gold salt complexes have been used. Though the mechanism is still uncovered, it is assumed that Au salts interact with albumin and are taken up by the immune cells, causing antimitochondrial effects and the apoptosis of cells [85–87]. Head and neck tumors showed specificity towards the Pt-based compounds; they might work by cross-linking the DNA in tumor cells [88]. For the treatment of manic-depressive disorder, lithium carbonate (Li_2CO_3) is used [89].

To prevent the contagiousness of infection, Ag has been used for various remedies since 4000 BC. The bactericidal effect of silver is well established, and topically, it is used to prevent infection of burned skin; it is also being used for ulcerations, bone prostheses, orthopedic surgery, catheters, heart devices, and surgical apparatus [90].

Diabetes, atherosclerosis, cancer, myocardial ischemia, pulmonary TB, asthma, Alzheimer's disease (AD), and Parkinson's disease (PD) are only a few of the many chronic diseases for which drug delivery vehicles have been extensively studied and shown to be effective. A number of these medicines, including Caelyx®, Abraxane®, Myocet®, Mepact®, Rapamune®, and Emend®, have been marketed for human use after positive results in preclinical and clinical testing. The potential of innovative therapeutic agents, such as peptides, nucleic acids (RNA and DNA), and genes, to be exploited as nanomedicines for the treatment of numerous chronic diseases, has been demonstrated beyond that of medications and chemicals [91].

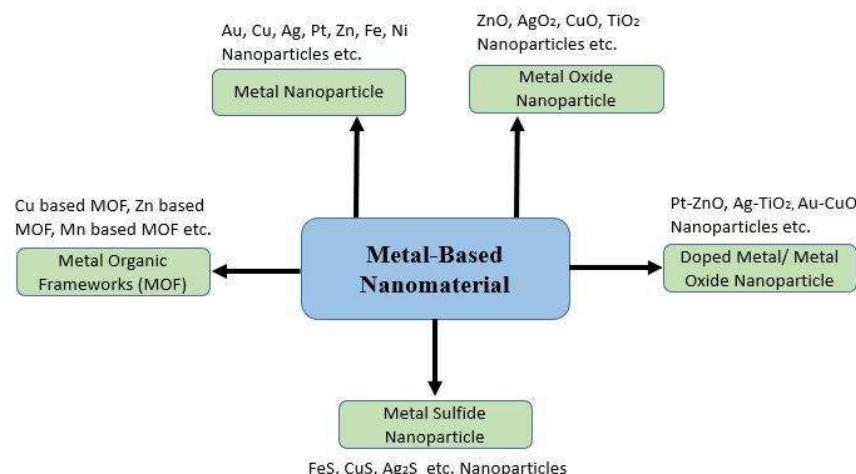


Figure 4. A couple of possible metallic nanoparticles. Adapted with permission from Ref. [91].

NPs based on metals such as Au, Ag, Fe, Cu, Pt, Zn, and so on have attracted a lot of interest in the medical field. NPs of metals have been demonstrated to exist in aqueous solutions, as demonstrated by Faraday [11]. Metallic NPs' hue and structure were analyzed by Kumar et al. [92] many years later. NPs can be manufactured and optimized in the present day by altering the chemical groups that aid in binding the antibodies. Ag-NPs could be utilized to treat a variety of skin ailments. Biomedical applications of noble metal NPs (Au, Ag, and Pt) include cancer treatment, drug transport, radiation therapy augmentation, thermal ablation, fungus elimination, diagnostic testing, and gene delivery, among many others. NPs of noble metals have special qualities that increase their worth. Peptides, antibodies, RNA, and DNA are just some of the functional groups that can be attached to metal NPs to make them more specific to the cells they are intended to target [93]. Some key NPs as well as their physiological applications are summarized below in Table 2.

Table 2. Some applications of metal nanoparticles.

Nanoparticles Name	Site of Action	Application	References
Au-NPs	Cancer glioblastoma-based multiforme	Radiosensitizer applications	[94,95]
Au-NPs	Cancer cell	Radiosensitizer application	[96]
Ag-NPs	Skin	Skin penetration evaluation	[97]
Pt-NPs lined with polyvinyl alcohol	Brain	Toxicity evaluation	[98]
Ag-NPs	Antimicrobial agent	Antimicrobial assessment	[99]
Au-NPs/Ag-NPs	Cancer cell	Photothermal therapy, imaging therapy	[100]
Au-branched shell nanostructure	Breast cell	Imaging therapy, photothermal therapy, chemotherapy.	[101]

1.6. Major Challenges of Using Nanoparticles in Medical Treatment

Firstly, though many testing procedures [102] have been developed for the evaluation of NP toxicity [103], these procedures are not universal for all NPs; they are designed for individual NPs and are not applicable for hybrid NPs. This fact leads to an undesirable outcome from the real objects and could be harmful for the body. As its effect is dependent on the size, shape, surface charging condition, and capping agents, it is really difficult to develop an accurate strategy to find out the toxicity. On the other hand, Au-NPs' effect also depends on its target receptor or organ; for example, different NPs show their effect at different concentrations. Thus, it is urgent to formulate some universal methods, such as good laboratory practice (GLP), to evaluate the safety of the NPs [103–105]. The adsorption of proteins to the particles also correlated with their physical characteristics (size, shape, charge, etc.) [106]. Significantly, metal oxide NPs have a high tendency to produce toxicity, this toxicity can be caused by a variety of mechanisms, including oxidative stress, coordination effects, nonhomeostatic effects, genotoxicity, and others. Size, solubility, and exposure routes all have an impact on metal oxide nanoparticles [107].

Secondly, although Au-NPs show outstanding result in tumor disease, there is a lack of studies to find out its pharmacokinetics (clearance and bio distribution) inside the human body. In vitro and in vivo studies cannot give the full picture of the biodistribution in the organism, which limits the wide use of gold NPs [108].

Third, this study discovered that in the case of tumor treatment, only 0.7% of NPs were able to reach cancer cells, with some exceptions reaching more than 5%. Moreover, when the NPs are injected in the blood circulation, they get absorbed in the mononuclear phagocytic system (MPS) and renal system, which reduces the effectiveness of the MPS day by day.

Finally, because there have been few Au-NP clinical trials, the data do not allow for comprehensive research on clearance, distribution, and protein absorption. Thus, a comprehensive trial for safety and toxicity should be carried out [109].

This review aims to highlight experiments conducted in the path of advancement in the therapeutic use of above discussed six metal NPs, such as Ag, Au, Zn, Cu, Fe, and Ni; additionally, we have used the literature to highlight the possible mechanism of action of significant effects of the selected potential metal NPs.

2. Therapeutic Applications of Metal NPs

2.1. Therapeutic Interventions of Gold Nanoparticles (Au-NPs)

When Robert Koch discovered that gold cyanide had a bacteriostatic effect on *Mycobacterium TB*, the medical use of gold for the treatment of tuberculosis was established for the first time. This led to the introduction of gold as a medicine in the 1920s [110].

Au-NPs have a tendency to aggregate at tumor sites [95]. Tumor cells can be killed by Au-NPs in a variety of ways, including as drug delivery systems for mechanical damage, anticancer medicines, and photothermal ablation [111].

In particular, Au-NPs are used in drug delivery, imaging, photo-thermal therapy, sensing, catalysis, and antimicrobials [112]. The list of applications of Au-NPs is much longer

because of their unique properties (Table 3). The biocompatibility of gold nanoparticles has been well documented; however, the typical reduction procedures used to create them can leave behind harmful chemical species [113]. Consequently, Au-NPs manufactured in an environmentally friendly manner hold far more promise in a variety of settings. Although Au-NPs are not as widely used as Ag-NPs as antibacterial agents, they nonetheless have considerable impact against a wide range of diseases due to their inherent biocidal qualities [112,114].

Au-NPs of 60 nm showed a positive result in retinoblastoma treatment [115], Au nanopopcorn 28 nm in size is used to diagnose prostate and breast cancer [116], and Au nanostars (Au-NS) 30 and 60 nm in size can be used to identify brain tumors, and this same NP showed a satisfying result against bladder cancer [117].

Silica-coated Au nanorods showed effective antitumor activity, both *in vivo* and *in vitro*, against breast cancer by targeting CD44+ receptors [118]. Colloidal Au-NPs are of interest as nontoxic carriers for drug delivery [119–121]. In a study, it was found that the internalization of the 50 nm spherical gold nanoparticles (AuNPs) was the best of all the nanoparticles investigated [122]. TrxR (thioredoxin reductase) function can be inhibited by gold compounds, which causes tumor cells to accumulate reactive oxygen species (ROS) and experience oxidative stress, which ultimately kills the tumor cells [123,124] and the proposed anticancer mechanism of Au-NPs is illustrated in Figure 5.

Nanotherapeutic Application of Gold

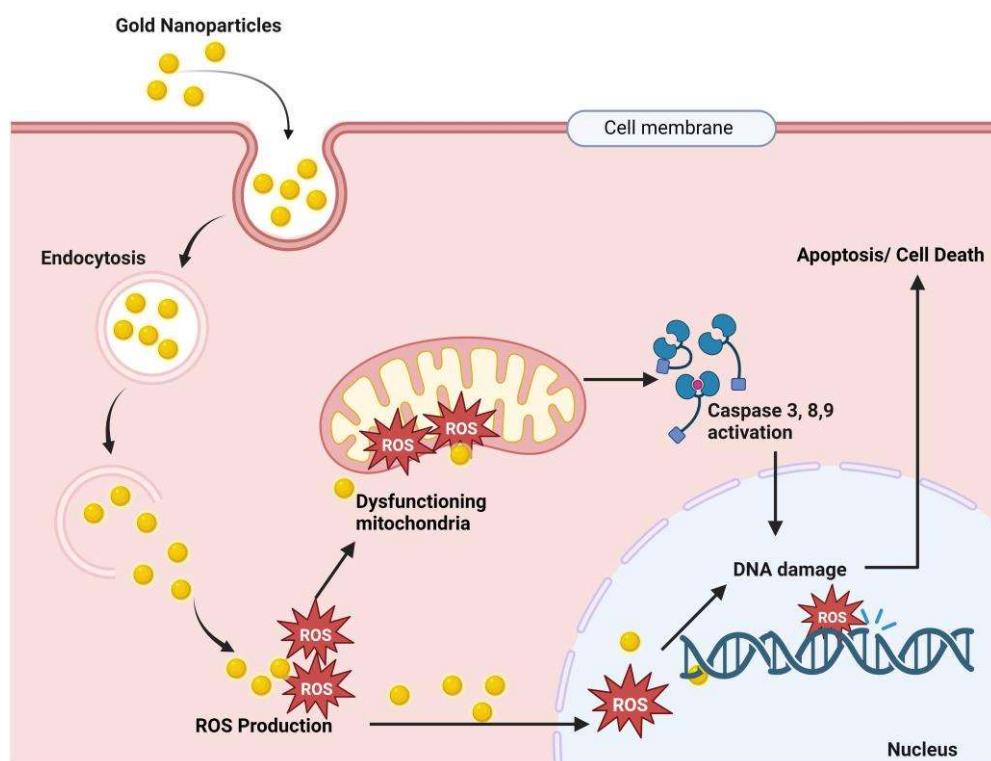


Figure 5. Proposed anticancer mechanism of gold nanoparticles. Here, Au-NPs pass through the cancer cell membrane by endocytosis, and endosomal release causes ROS (reactive oxygen species) production. These ROS cause mitochondrial dysfunction and result in caspase 3, 9, and 8 activations, which results in DNA damage and finally cell death [123–126].

Table 3. Nanotherapy of gold nanoparticles.

Nanoparticles (Diameter)	Test Medium	Concentration	Effect/Result	Disease Against	References
<i>Anticancer Effect</i>					
Au-nanopopcorn (28 nm)	LNCaP (prostate cancer cells)	0.5 mL	Popcorn-shaped Au-NPs enhanced Raman intensity to recognize prostate cancer cells.	Prostate and breast cancer	[105]
Au-NPs	Y79 (MTT analysis)	1.75, 3.5, 7, 14, 28 and 56 μ g/mL	Forty-eight hours after applying Au-NPs, 0.5 to 11 min hyperthermia is applied, which shows 50% cell viability after 4.5 min, but without NPs, 9 min is required to obtain same effect.	Hyperthermia in cancer	[115]
Silica coated Au nanorods	MB49 bladder cancer cell line in mice	0.1 nM	Synergistic immuno photo nano therapy (SYMPHONY) produce better survival than other groups. In photo thermal monotherapy Au-NPs has much efficiency than nanoshells.	Bladder cancer	[117]
DOX@ Au-NPs (2 nm Au-NPs)	Mammary carcinoma cells	-	Showed efficient in vivo and in vitro antitumor activity in targeting CD44+ receptor.	Breast cancer	[118]
PDC-PEG-Au-NPs (25–50 nm GNPs)	Breast Cancer Cell lines (MCF-7 and MDA-MB-231); Murine Mammary 4T1; CD-1 Mice	5 mg/kg dose	Good renal clearance with fruitful targeting. Decreased normal tissue toxicity with improved antitumor efficacy.	Breast cancer	[119]
BLM-DOX-PEG-Au-NPs (13 nm GNPs)	Murine lymphoma cells (A20)	Up to 50 μ M dose	Half-life of drug increased, toxicity observed in targeted cells, and effective for a long time.	Articancer	[120]
Au-nanostars (Au-NS) (30 and 60 nm)	HeLa, cervical cancer cell line	10–100 nM dose	Cancer cell environment-mediated drug release and improve EC ₅₀	Cervical cancer	[121]
Chitosan/Au-NPs	Glioblastoma model on mice	0.1 nM	7.2% ID/g uptake of Au-NPs in the brain tumor, which is identified by PET/CT scan.	Brain tumor	[127]
NP-based nucleic acid conjugates Au@GO NP-NACs	HepG-2 and Caco-2 cell lines	0.1, 0.05, 0.025, 0.0015 mg/mL	Cancer cell proliferation is inhibited more than chitosan.	Cancer and bacterial infection	[128]
	Result identified by in vitro microfluidic models, Bcl-2	100 μ g/mL of Au-NPs	Better live cancer cell identification by SERS and synergistic and specific killing of cancer cells.	Articancer	[129]

Table 3. *Cont.*

Nanoparticles (Diameter)	Test Medium	Concentration	Effect/Result	Disease Against	References
Cancer-targeting peptide-functionalized NP (3.52 nm) using Au-NPs (26.2 nm) and TA-peptide complex	MCF-7 and T47D (breast cancer cell lines), on tumor containing mice	-	Targeted cell death by apoptosis. Considerable hemocompatibility, higher release of cytochrome c, and higher antitumor activity is found.	Breast cancer	[130]
Chitosan coated Au-nanospheres		IC ₅₀ value 127 μ M	This study compares the cellular uptake of Au-NPs and found that Stars < rods < triangles (lowest to highest uptake order). The mechanism of cellular uptake was endocytosis.		
Chitosan coated Au-NS	RAW264.7 cells	IC ₅₀ value 81.8 μ M	Au-NP nanorods showed greater cellular uptake and high cytotoxicity against RAW264.7 cells.	Anticancer	[131,132]
Chitosan-coated Au-nanorods		IC ₅₀ value 22.7 μ M			
Au-NPs	Osteosarcoma mouse model	-	Au-NPs with CD133 and hyaluronic acid increased the photo thermal antitumor therapy. HA can gourd the photosensitive drugs from photo-degradation and inhibited the proliferation of osteosarcoma cells.	Bone cancer	[133]
Au-NPs covered with multivalent hydrocarbon. (6.9 \pm 2.9 nm)	Xenograft mouse model with the HeLa, SCC7, and SKBR3 cancer cell lines	100 nM–10 mM	Tumor growth was considerably suppressed in C18/F127 injected in xenograft mice compared with the control group.	Breast cancer	[134]
Anti-HER2 functionalized Au-on-silica nanoshells	RAW 264.7 cells	LD ₅₀ : 1 mg /mL and 10 μ g /mL for cationic carrier.	Targeted action; precisely eliminating cancer cells while protecting healthy tissues	Breast cancer	[135]
Au-NPs produced from <i>Enterococcus</i> sp.	Colorectal tumor cells (HT-29)	5–24 μ g /mL	Inducing ROS and caspase-3 expression, weakening the potential of mitochondrial membrane.	Anticancer	[136]
PTX-TNF α -PEG-Au-NPs (32.6 nm)	Ovarian cancer cell line (A2780); B16/F10 tumor induce C57BL/6 mice	2.5 mg/kg dose	Specific delivery of NPs to tumor and improved efficacy	Ovarian cancer	[137]
DOX-PEG-Au-NPs (41 nm Au-NPs)	Ovarian cancer cell line (A2780); CD-1 mice	6 mg/kg dose	Significantly reduced normal tissue toxicity	Ovarian cancer	[138]
CIS-GLC-PEG-Au-NP (20 nm GNPs)	Skin cancer cell line A-431; A-431 cell line bearing mice	10 mg/kg dose; 25 Gy at 6 MV	Same type effect to free cisplatin; improved result when used in combination with radiation	Skin cancer	[139]

Table 3. *Cont.*

Nanoparticles (Diameter)	Test Medium	Concentration	Effect/Result	Disease Against	References
Alginate conjugate with Au-NPs and CIS (44 nm NP)	Cervical cancer cell line (KB)	20 µg/mL dose of Au-NP along with 5 µg/mL CIS; 4 Gy at 6 MV	ACA and radiotherapy found increased efficacy over cisplatin and radiation. Using photothermal therapy further enhanced the anticancer effect.	Cervical cancer	[140]
5-FU/GSH-GNPs (9–17 nm Au-NPs)	Colorectal cancer cell lines (isolated from patients)	0.5–1.5 mg/mL dose	Better anticancer effect, and minimized the drug doses as a result.	Colorectal cancer	[141]
Cs-Au-NPs-DOX (21 nm GNPs)	MCF-7, Breast cancer cell line	0.05–0.3 mM dose; 0.5, 1, and 3 Gy at 6 MV	Improved test results, decreased survival fraction, upregulated apoptosis, and DNA damage.	Breast cancer	[142]
Au-NP-PEG-RGD; CIS (10 nm Au-NPs with 435 nM CIS)	MDA-MB-231, Breast Cancer Cell line	0.3 nM dose; 2 Gy at 6 MV	Increased efficacy of treatment compared to cisplatin or radiation alone.	Breast cancer	[143]
Au-NP-PEG-RGD; DTX (17.2 nm GNPs)	Breast Cancer Cell line (MDA-MB-231) and Cervical Cancer Cell line (HeLa)	0.2 nM Au-NPs with 50 nM DTX; 2 Gy at 6 MV	Greater retention of Au-NPs due to cell synchronicity induced by DTX. Synergistic therapeutic action observed when Au-NPs and DTX were combined.	Breast cancer	[144]
Antimicrobial Effects					
Gold-chitosan hybrid NPs (16.9 nm)	Tested against the <i>S. aureus</i> (Gram-positive) <i>P. aeruginosa</i> (Gram-negative) bacteria	0.25 mg/mL	The action is still not clear.	Bacterial infection	[145]
Au-NPs (17 nm)	HIV-1	0.05–0.12 mg/mL	Au-NPs inhibits HIV-1 but its mechanism is unknown.	Viral infection	[146]
Au-NPs (25 nm)	<i>Candida</i> sp	16–32 µg/mL	Cell death for intracellular acidification by the inhibition of H ⁺ ATPase.	Fungal infection	[147]
IgG-Au-NPs (32 nm)	MRSA cultures	1–50 mg/L	6.25% minimum inhibition concentration (MIC) for the IgG-Au-NPs, while 25% MIC was found for Au-NPs alone.	Methicillin-resistant <i>Staphylococcus aureus</i> (MRSA) infection	[148]

Table 3. *Cont.*

Nanoparticles (Diameter)	Test Medium	Concentration	Effect/Result	Disease Against	References
<i>Miscellaneous Effects</i>					
Au-NPs/chalcones conjugate (2 to 12 nm)	HEK293 cells	20–100 µg/mL	Therapeutic development of antidiabetic drug, which is derived from <i>H. foetidum</i> by increasing glucose uptake and no particle shows cytotoxicity against HaCaT keratinocytes. Helichrysetin is a potential compound for antidiabetic effect.	Antidiabetic	[149]
Au-NPs/chalcones conjugate (2 to 12 nm)	α-amylase and α-glucosidase enzyme	20–100 µg/mL	Potential enzyme inhibitory activities against α-amylase and α-glucosidase enzymes.	Enzyme inhibition	[149]
Au-NPs	pBR322 (plasmid DNA)	64 ng/mL (nanogram)	γ-ray radiation applied by HDR brachytherapy, ROS (reactive oxygen species) formation and DNA breaks occurred in positive charged Au-NPs but not in negative charged Au-NPs.	Plasmid DNA damage	[150]
SPIO-Au-NPs (FeO-Au) core-shell NPs	PC-12 cells (Neuron like cell)	127 µg of SPION-Au-NPs	Shows higher intracellular interaction with PC-12 neuron-like cells.	Neuroregeneration	[151]
Synergistic Immuno Photothermal Nanotherapy (SYMPHONY)	Tumor cell treated 30, 60 nm	0.05 nM NPs with radiation	In murine animal models, it provides a ‘cancer vaccine’ effect that leads to immunologic memory and inhibits cancer recurrence.	Photoimmunotherapy	[152]
Peptide-coated Au-NPs	Human peripheral blood mononuclear cells	12.5–50 µg/mL	Efficiently suppressed TLR signaling and shielded mice from LPS-induced acute lung injury. PPIs and the recently found that Au-NPs-based TLR inhibitors have comparable modes of action.	Acute Lung Injury	[153]

2.2. Therapeutic Interventions of Silver Nanoparticles (Ag-NPs)

Silver has excellent physicochemical features, such as catalytic, optical, electric, and, of course, antibacterial capabilities, and these qualities make silver nanoparticles the most marketable nanoparticles. In the presence of Ag-NPs, the synergistic impact of antibiotics such as cefotaxime, azithromycin, cefuroxime, chloramphenicol, and fosfomycin against *E. coli* was greatly boosted as compared to antibiotics alone [80].

Other metal NPs may exhibit equivalent efficacy against particular germs, but overall, silver is said to be the most effective material against a variety of pathogens. Ag-NPs inhibit the extracellular activity of severe acute respiratory distress syndrome coronavirus 2 (SARS-CoV-2) [154].

Ag-NPs are the preferred metal when antibacterial characteristics are required. The antibacterial, antiviral, antioxidant, and anticancer characteristics of silver are well recognized, and it has the potential to be developed into a unique therapeutic agent. Ag also has antiparasitic, antiviral, and anticancer qualities [155,156], and the mechanisms of action of these effects are illustrated in Figure 6. Ag-NPs, after entering cells by endocytosis, produce ROS that damage the endoplasmic reticulum and mitochondria. The cellular pathways NF- κ B, PI3K/AKT/mTOR, Wnt/beta-catenin, MAPK/ERK, and ERK activation result in DNA fragmentation, cell cycle arrest, and cell apoptosis [157–161]. Table 4 shows the prominent nanotherapeutic applications of silver.

Nanotherapeutic Application of Silver

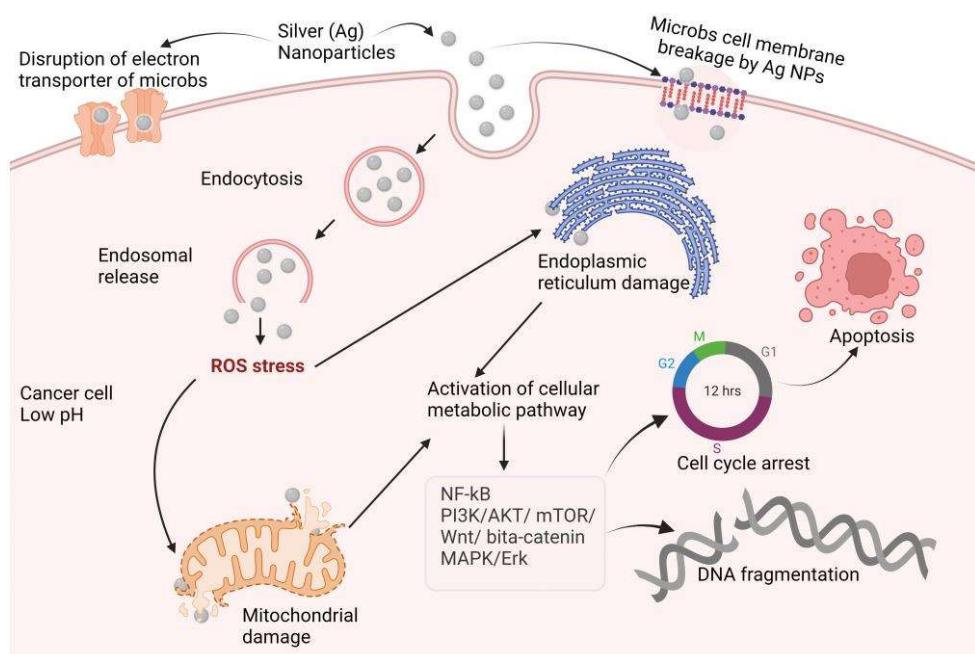


Figure 6. Proposed anticancer mechanisms of silver nanoparticles (NF- κ B: nuclear factor kappa-light-chain-enhancer of activated B cells; PI3K: phosphoinositide 3-kinases AKT: protein kinase B; mTOR: mammalian target of rapamycin; Wnt: wingless and Int-1; MAPK: mitogen-activated protein kinase; ERK: extra-cellular receptor kinase).

Table 4. Nanotherapy of some silver nanoparticles.

Nanoparticles (Diameter)	Test Medium	Concentration	Effect/Result	Disease Against	References
<i>Anticancer Effects</i>					
Ag-NPs (5–20 nm)	MCF7-FLV cell line	136 µM	Cytotoxic effects against breast cancer.	Breast Cancer	[70]
Ag-NPs (26.18 nm)	Human alveolar cancer cell line: A549	87 and 41 µg/mL	Activity against the A549 cell line without showing any damage in noncancer cells.	Alveolar cancer	[162]
Ag-NPs (50–70 nm)	Human acute T cell leukemia cell line	10 to 50 µM	Cytotoxic activity against leukemia.	Leukemia	[163]
Ag-NPs (33 nm)	Human cervical cancer cells (HeLa)	10 to 50 µg/mL	Induced cytotoxicity in HeLa cells in a concentration-dependent manner.	Cervical cancer	[164]
Ag-NPs (24–150 nm)	HCT-116 cells colon cancer cell	100 µg/mL	The sub-G1 phases of the cell cycle were changed, and larger levels of fragmented DNA were discovered.	Colon cancer	[165]
Ag-NPs (6 nm) with gemcitabine (GEM)	Human ovarian cancer cell line A2780	50% inhibitory concentration (IC_{50}) of GEM and Ag-NPs after a 24 h exposure was 100 and 90 nM	Lowering cell viability and proliferation, as well as increasing LDH leakage and ROS production.	Ovarian cancer	[166]
Ag-NPs (2.8 and 18 nm)	PANC-1 and hTERT-HPNE	1.67 µg/mL for 2.8 nm size and 26.81 µg/mL for 18 nm size	Ag-NPs triggered programmed cell death in PANC-1 cells, including apoptosis and necrosis, as well as autophagy and mitotic catastrophe, in a concentration- and size-dependent manner.	Pancreatic ductal adenocarcinoma	[167]
Ag-NPs (60 nm)	MG63 osteosarcoma cell line	81.8 ± 2.6 and 75.5 ± 2.4 µg/mL	Chromatin condensation causes dose-dependent cytotoxicity and ultimately cell death.	Osteosarcoma	[168]
Ag-NPs (lesser than 50 nm)	Pleomorphic hepatocellular carcinoma (SNU-387), hepatic ductal carcinoma (LMH/2A), morris hepatoma (McA-RH7777), and novikoff hepatoma (N1-S1 Fudr) cell lines	477, 548, and 605 µg/mL	In the presence of Ag nanoparticles, the liver malignant cells viability decreased.	Liver cancer	[169]
Ag-NPs (30 to 90 nm)	A431 human skin cancer cells	64.2 µg/mL	Showed a high level of cytotoxicity against the A431 cell line.	Skin cancer	[170]

Table 4. *Cont.*

Nanoparticles (Diameter)	Test Medium	Concentration	Effect/Result	Disease Against	References
Ag-Cys-NPs	Glioma and neuroblastoma cells	100 and 1000 ng/mL	Ag-Cys-NPs is about 10-fold potent than the Cu-NPs for SH-EP 1 cells and Ag-Cys-NPs is 20 folds more potent than Cu-NPs for glioma cells.	Anticancer	[171]
<i>Antibacterial Effects</i>					
Ag-NPs (10 nm)	<i>Vibrio cholerae</i>	40 µg/mL	Antibacterial efficacy of microbial GLP-capped Ag-NPs against <i>V. cholerae</i> .	Cholera	[172]
Ag-NPs (70 nm)	<i>Mycobacterium tuberculosis</i> , H37Rv	6.25–50 mM	Mild growth-inhibitory effect.	Tuberculosis	[173]
Ag-NPs (12.62–27.45 nm) with imipenem	<i>Klebsiella pneumoniae</i> clinical strain	Concentration below 3 mg/L	The antibacterial properties of AgNPs in conjunction with imipenem were extended against IRKP infection.	Pneumoniae	[174]
<i>Antiviral Effects</i>					
Ag-NPs (10 nm)	SARS-CoV2	1–10 ppm	Inhibiting extracellular activity of SARS-CoV2.	COVID-19	[154]
Ag-NPs (30–50 nm)	HeLa-CD4-LTR-β-gal cells, MT-2 cells, human PBMC	3.9 ± 1.6 mg/mL against HeLa-CD4-LTR-β-gal cells, as 1.11 ± 0.32 mg/mL applied against human PBMC, and 1.3 ± 0.58 mg/mL used against MT-2 cells.	HIV particles are turned inactive quickly, allowing for early disruption of the viral replication cycle.	HIV	[155]
Ag-NPs (3.5, 6.5, 12.9 nm)/Ch composite	H1N1 influenza A virus suspension	250 µL Ag NP/Ch composite	Antiviral activity against H1N1 influenza. Provides a concentration-dependent effect.	Influenza	[175]
Ag-NPs (13, 33 and 46 nm)	HSV-1 and HSV-2	2.5 µL	Vero cell infection by HSV-1 and HSV-2 is downregulated in a dose-dependent manner.	Herpes	[176]
Ag-NPs (10 nm)	HepAD38 cell line	5 to 50 µM	Suppressing HBV RNA and extracellular virions generation in vitro.	Hepatitis B	[177]
Ag-NPs (70–95 nm)	<i>Chikungunya virus</i> (CHIKV)	31.25 µg/mL	By inhibiting the cytopathic impact, showing excellent efficacy against CHIKV.	Chikungunya	[178]
Ag-NPs (100 nm)	Serotype DEN-2	20 µL/mL	Plaque assay estimates of dengue virus output were lowered.	Dengue	[179]

Table 4. Cont.

Nanoparticles (Diameter)	Test Medium	Concentration	Effect/Result	Disease Against	References
<i>Miscellaneous Effects</i>					
Ag-NPs (37 nm)	<i>Propionibacterium acnes</i>	3.1 µg/mL	The mechanism of silver colloid particles bactericidal action on bacteria is still being investigated.	Acne	[180]
Ag-NPs (37 nm)	<i>Malassezia furfur</i>	25 µg/mL	Antifungal activity was highest against <i>M. furfur</i> .	Dandruff	[180]
Ag-NPs (53 nm)	α-amylase and α-glucosidase	54.56 and 37.86 mg/mL	Inhibition of carbohydrate digestion enzymes, for example α-amylase and α-glucosidase, was effective.	Diabetes	[181]
FA-Ag-NPs	Murine macrophage cells (RAW264.7), mice, age: 7–8 weeks.	0.652 nmol/kg	Rheumatoid arthritis treatment was performed by simultaneously M1 macrophage apoptosis and M1-to-M2 macrophage re-polarization.	Rheumatoid arthritis	[182]
Ag-NPs-PADM hydrogel (PADM = porcine dermal extracellular matrix), 5 and 50 nm	Rat, age: 6-month, weight: 200–300 g	20, 50, and 80 µg/mL	In vivo, Ag-NPs-PADM hydrogel enhanced angiogenesis and repaired infected skin defects.	Skin infection defect	[183]
Ag-NPs 20–35 nm	Rat model	Orally administered Ag-NPs concentrations of 175 and 350 ppm	Ag NPs have a gastroprotective effect in rats against ethanol-induced gastric ulcer. Superoxide dismutase (SOD) and catalase (CAT) activities were increased by Ag NPs.	Gastroprotective	[184]

2.3. Therapeutic Interventions of Copper Nanoparticles (Cu-NPs)

Researchers and health care professionals have been drawn to cupric oxide (CuO) NPs for their physical, chemical, high temperature, and photocatalytic capabilities, but most notably for their antibacterial properties [185]. Copper nanoparticles' synergistic activity with amoxicillin, ampicillin, ciprofloxacin, and gentamicin against both Gram-positive and Gram-negative bacteria was investigated, and ampicillin showed comparatively improved activity compared to alone [186]. Cu-NPs inactivate glycosidase to provide an antidiabetic effect, and the study found that Cu-NPs showed an anticancer effect by activating BAX and p53 and by decreasing Bcl-2 expression, which result in apoptosis in cancer [187]. Cu-NPs increase ROS production in bacterial cells and cause bacterial DNA and protein destruction; on the other hand, accumulation of Cu-NPs in the bacterial cell wall causes cell wall disruption [188–194].

The mechanisms underlying these effects are depicted in Figure 7. Other nanotherapeutic applications of copper are presented in Table 5.

Nanotherapeutic Application of Copper

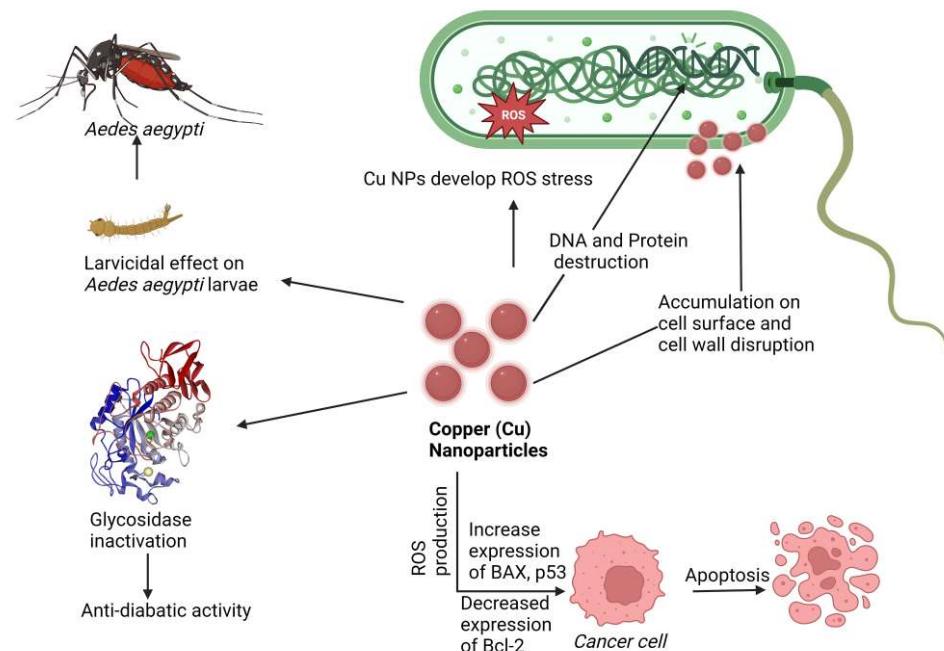


Figure 7. Proposed mechanism of nanotherapeutic applications of copper. Here Cu-NPs showed an anticancer effect by increasing BAX and p53 expression and Bcl-2 downregulating, an antidiabetic effect by glycosidase inactivation, an antimicrobial effect by ROS production cell wall disruption, and a larvicidal effect against *Aedes aegypti* (Dengue virus carrier).

Table 5. Nanotherapy of copper nanoparticles.

Nanoparticles (Diameter)	Test Medium	Concentration	Effect/Result	Disease Against	References
<i>Anticancer Effect</i>					
Cu-NPs/CS-Starch (5–7 nm)	TPC1, BCPAP and FTC133	207 µg/mL	TPC1, BCPAP, and FTC133 cell lines shown substantial antithyroid activity.	Thyroid cancer	[95]
Cu-NPs (62.7 nm) with albumin	MDA-MB 231 cell line	70 µM	Suppressed cancer cell viability while being less harmful to normal cells.	Breast cancer	[190]
Cu-NPs (4.7 to 17.4 nm)	HepG2 cells	19.88 µg	HepG2 cells have a high cytotoxic activity.	Hepatic cancer	[193]
Cu-NPs (4.7 to 17.4 nm)	Caco-2 cells	11.21 µg	Inhibition of Caco-2 cell growth.	Colon cancer	[193]
Cu-NPs (10–20 nm) with chitosan	UM-UC-3 (Transitional cell carcinoma), SCaBER (Squamous cell carcinoma), and TCCSUP (Grade IV, transitional cell carcinoma)	238, 404, and 569 µg/mL	Cytotoxic activity against common bladder cancer cell lines in humans.	Bladder cancer	[195]
Cu-NPs (39.3 ± 5.45 nm)	Human skin carcinoma cells (B16F10) and mouse embryonic fibroblast cell line (NIH3T3)	40 and 120 µg/mL	Mice showed potential suppression of B16F10 melanoma cell proliferation and tumor development inhibition.	Melanoma	[196]
Cu-NPs (12–16 nm)	Human lung carcinoma cells (A549)	20–100 µg/mL	In a dose-dependent way, lung cancer cells showed extensive structural damages and increased oxidative stress indicators.	Lung carcinoma	[197]
Binmetallic CuFe (copper–iron) PBA and CoFe (cobalt–iron) PBA NPs	Tumor tissues for in vitro and BALB/c mice for in vivo test	5, 10, 20, 40, 80, and 160 µg/mL	Prussian blue analogs (PBA-DDSs) prepared with metal NPs doxorubicin (DOX) delivery and pH-controlled release development.	Breast cancer	[198]
Cu-NPs (15 ± 1.7 nm)	HeLa, A549, and BHK21 cell lines	120 µM	Caused the death of tumor/ cancer cells through apoptosis.	Artitumor	[199]
Cu-NPs with chitosan (<20 nm)	CHO cells and MC3T3-E1 preosteoblast cells	1–1000 µg/mL	A higher degree of mitochondrial ROS production.	Osteosarcoma	[200]
CuHARS	Cell line of a glioma tumor	20 µg/mL	CuHARS decreases the glioma cell and BMVECs viability 20% and 200% respectively. Immune supportive by the production of NO.	Antitumor and immunomodulatory	[201]

Table 5. *Cont.*

Nanoparticles (Diameter)	Test Medium	Concentration	Effect/Result	Disease Against	References
<i>Antiviral Effects</i>					
Cu-NPs (20 nm)	SARS-CoV-2	500 µL	By putting virus-containing media onto copper-coated PP filters and then adding Vero cells, inactivation was assessed.	COVID-19	[189]
Cu-NPs (13.5 ± 0.6 nm)	<i>Culex quinquefasciatus</i> , <i>Anopheles stephensi</i> , and <i>Aedes aegypti</i>	500 µg/mL	A mortality rate that was dosage and time dependent.	Chikungunya	[191]
<i>Miscellaneous Effects</i>					
Cu-Epigallocatechin-3-gallate (Cu-EGCG)	<i>Aedes aegypti</i> larvae	55.12 mg/mL	Assessing the larvicidal efficacy of <i>Aedes aegypti</i> .	Dengue	[192]
Cu-NPs (30 and 50 nm)	Female Sprague rats	50, 100, 200 µg/mL	Inhibited bacteria such as <i>E. coli</i> and <i>S. aureus</i> to protect from wound infection.	Wound healing	[56]
Cu-NPs (12–16 nm)	<i>Streptomyces griseus</i>	-	Nanocopper has the potential to be an effective new fungicide.	Red root-rot disease	[185]
Cu-NPs (100 nm)	<i>Fusarium equiseti</i> F. <i>oxysporum</i> and F. <i>cultorum</i>	10 µg/mL	Inhibitory actions of glycosidase in vitro.	Antidiabetic	[188]
Cu-NPs (spherical 2.88 ± 0.94, triangular 1.27 ± 0.37 and hexagonal 1.81 ± 0.52 nm)	Cultured porcine ovarian granulosa cells	25, 20 and 19 mm	Exhibited antifungal efficacy against F. <i>oxysporum</i> .	Crop diseases	[194]
Cu-NPs (17 and 41 nm)	<i>T. gondii</i> tissue cysts	1, 10, or 100 ng/mL	The ability to influence viability, proliferation, apoptosis, and the release of steroid hormones.	Reproductive disorders	[202]
CuHARS (polymer-coated copper cystine high-aspect ratio structures); 60–100 nm	<i>Escherichia coli</i> and <i>Staphylococcus epidermidis</i>	0.2 and 0.3 mL/kg and in combined with atovaquone (100 mg/kg)	Infected mice with <i>T. gondii</i> had substantial prophylactic effects when combined with atovaquone.	Toxoplasmosis	[203]
			NO production facilitates antimicrobial action of CuHARS.	Antibacterial	[204]

Table 5. *Cont.*

Nanoparticles (Diameter)	Test Medium	Concentration	Effect/Result	Disease Against	References
CuO.MBGs; Mesophorous bioactive glasses (MBGs) (10–20 nm)	In vitro simulated body fluid (SBF)	5% of CuO NPs in MBG	Outstanding biomaterial for bone regeneration. MBGs released therapeutic amounts of Ca ²⁺ and Cu ²⁺ ions.	Bone defect	[205]
CuS incorporated hyaluronic acid (injectable hydrogel); average 35 nm	SD male rats; weight range 200~220 g	200, 100, 50, 20, and 10 µg/mL	Improved wound healing and angiogenesis occur.	Wound healing	[206]

2.4. Therapeutic Interventions of Zinc Nanoparticles (Zn-NPs)

Zinc is a material that is frequently used in biomedical applications due to its unique features, such as electric conductivity, optical capabilities, and piezoelectric qualities [207]. Beyth et al. defined the method of killing bacteria using zinc oxide (ZnO) NPs as having two pathways of action [208]. The first involves cell wall penetration, and the second includes the formation of ROS. Zn-NPs follow the Bcl-2/BAX/BAK pathway to cell apoptosis by caspase-3 and -9 and ROS-induced DNA fragmentation leading to cell cycle arrest and apoptosis, and also follow the mitochondrial disruption for an anticancer effect [209–211], as shown in Figure 8.

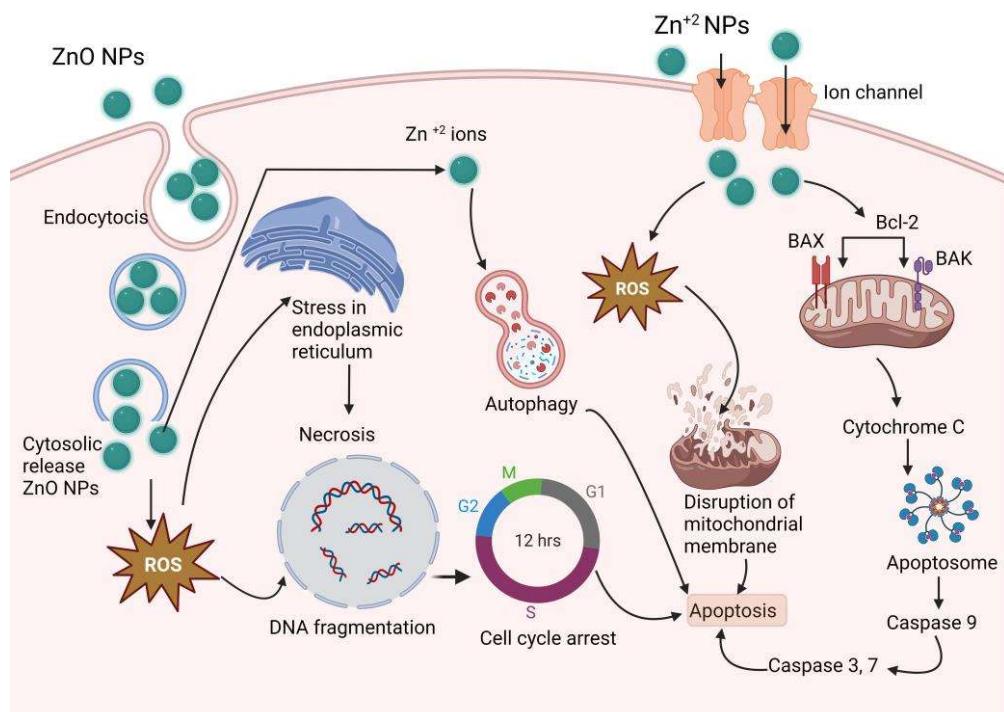


Figure 8. Proposed anticancer mechanism of Zn-NPs (ZnO-NPs create stress in endoplasmic reticulum, and produce ROS, which results DNA fragmentation and cell cycle arrest; on the other hand, produced ROS disrupts mitochondrial membrane and activates caspase 3, 7, and 9, which results in apoptosis).

ZnO-NPs have antibacterial, antifungal, anticancer, antidiabetic, and antitubercular activity, and breast cancer inhibition is an optimistic property that this study observed in a number of studies (presented in Table 6). Even 100 nm Zn-NPs supplemented at 30 ppm improved growth and serum glucose levels in layer chicks [212].

Nanotherapeutic Application of Zinc

Table 6. Nanotherapy of some zinc nanoparticles.

Nanoparticles (Diameter)	Test Medium	Concentration	Effect/Result	Disease Against	References
<i>Anticancer Effects</i>					
ZnO-NPs (16–19 nm)	Breast cancer cell (MCF7), and Lung Cancer cell (A549)	31.2 µg/mL	The cell viability is reduced by NPs, which induces cytotoxicity in cancerous cells.	Anticancer	[213]
ZnO-NPs (100 nm)	Human Breast Cancer (MCF-7) cells	10 µg/mL	Apoptosis is provoked and induced through an intrinsic mitochondrial pathway, depending on caspase activation.	Anticancer	[214]
ZnO (36.91 ± 1.21 nm), ZnO@Ce6 (47.75 ± 0.05 nm) and ZnO@Ce6-PDA (51.92 ± 1.96 nm)	HeLa cells	30 µg/mL	Photothermal and photodynamic action, and increased the cell viability by more than 90%.	Anticancer	[215]
ZnO-NPs (30.4–40.8 nm)	MCF-7 cell line; (Breast cancer cell line)	25 µg/mL	The growth of Gram-positive <i>Bacillus licheniformis</i> is inhibited, reducing the viability of MCF-7 cells.	Breast cancer	[216]
ZnO-NPs (30.4–40.8 nm)	MCF-7 cell line; (Breast cancer cell line)	25 µg/mL	The growth of Gram-positive <i>Bacillus licheniformis</i> is inhibited, reducing the viability of MCF-7 cells.	Breast cancer	[216]
ZnO-NPs (66.25 nm)	MDA-MB 231 and MCF-7 breast cancer cell lines.	0.1, 0.05 and 0.01 M	The activity of MDA-MB 231 cells is inhibited with increased concentration.	Breast cancer	[217]
ZnO-NPs (10–15 nm)	MCF-7 cell lines	15.88 µg/mL	Inducing apoptosis in MCF-7 cell line via the Caspase-8 and p53 Pathway. Cancer cells may develop and spread throughout the body as a result of mutations (changes) in the p53 gene.	Breast cancer	[218]
PBA-ZnO (<40 nm)	MCF-7 cell lines	35 and 50 µg/mL	Cell death by apoptosis was induced in the MCF-7 cell line by enhancing oxidative stress and mitochondrial damage.	Breast cancer	[219]
ZnO-NPs (10–70 nm)	MCF-7 cell lines	50 µg/mL	Inhibiting apoptosis.	Breast cancer	[220]
Zn-Fe ₂ O ₄ -NPs (17.12 nm)	MCF-7 cell lines	25–500 µg/mL	Decrease in cell viability by cytotoxic activity.	Breast cancer	[221]

Table 6. *Cont.*

Nanoparticles (Diameter)	Test Medium	Concentration	Effect/Result	Disease Against	References
ZnO-NPs (31.5 nm)	MCF-7, MDA-MB-231, and HFF cell lines.	11.16 µg/mL	Reduction of the expression of micro-RNAs.	Breast cancer	[222]
Triton-X modified ZnO-NPs (13.45 ± 1.42 nm)	Breast cancer cell line (MDA-MB-231) and normal cell line (NIH 3T3) were used.	55.24 µg/mL	Cytotoxicity is enhanced through surface modification.	Breast cancer	[223]
PEG-ZnO-NPs (150 nm)	DMEM medium (HiMedia)	6.25–37.5 µg/mL	The impairment of DNA damage repair enzyme NEIL2 by inducing apoptosis in breast cancer cells through ROS.	Anticancer	[224]
MSN-ZnO-Au-NPs (76.5 ± 11.8 nm)	Breast cancer cells (MCF-7: estrogen receptor-positive, CA151: triple-negative).	25 µg/mL	The viability of all cell lines is reduced.	Resistant breast cancer	[225]
ZnO-NPs (12–14 nm)	MDA-MB-231 cancer cells.	7.103 µg/mL	Decreasing cell viability by cytotoxic impact.	Breast cancer	[226]
ZnONPs (25–40 nm)	Michigan Cancer Foundation-7 [MCF7], and murine (TUBO) breast cancer cell lines	8, 4, and 2 µg/mL	Inducing apoptosis by increasing the concentration of ZnO-NPs.	Antitumor	[227]
Zn-NPs (9–17 nm)	MCF-7 (breast carcinoma cell line), HCT-116 (colon carcinoma cells)	3.9, 7.8, 15.6, 31.25, 62.5, 125, 250 and 500 µg/mL were used.	For MCF-7, concentrations of 373 µg/mL and >500 µg/mL and for HCT-116, concentrations of 226 and 317 µg/mL were found effective in the in vitro test.	Antitumor	[228]
<i>Antibacterial Effects</i>					
ZnO-NPs (2–28 nm)	<i>Pseudomonas</i> sp., <i>Fusarium</i> sp.	0.1 M	Bacterial membranes are disrupted by the formation of ROS, for example superoxide and hydroxyl radicals.	Bacterial and fungal infection	[229]
ZnO-NPs (45–150 nm)	<i>Helicobacter pylori</i> and human mesenchymal stem cells (hMSC)	3.125–100 µg/mL	Biocompatibility to hMSC and described as safe in mammalian cells and can be used as antibiotics.	Antibacterial	[230]
RF-contained Zn ²⁺ ion-cross-linked SA-g-AA-M PNNs. <300 nm	Vero cells	100 µg/mL	The gene transcription is inhibited by inhibiting the β-subunit of the bacterial RNA polymerase.	Tuberculosis	[231]

Table 6. *Cont.*

Nanoparticles (Diameter)	Test Medium	Concentration	Effect/Result	Disease Against	References
ZnO-NPs (125 nm)	<i>Streptococcus mutans</i>	3.90–4000 µg/mL	Bacteriostatic and bactericidal effects.	Microbial infection	[232]
ZnO-NPs (12 nm)	<i>Staphylococcus aureus</i> and <i>Escherichia coli</i>	5.6 µg/mL	The ROS production was increased, while the cellular function and cell membrane were disrupted.	Bacterial infection	[233]
MgZnO-NPs/PU (52.65 ± 2.58 nm)	<i>E. coli</i> , DH5α strain	9 × 10 ⁻⁵ CFU/mL	The damage of the structure and function of cell originals (mesosome), consequently affecting the deoxyribonucleic acid (DNA) replication by promoting ROS.	Bacterial infection	[234]
ZnO-NPs (20–45 nm)	Ciprofloxacin	500, 1000, and 2000 µg/mL	Increasing antimicrobial activity of ciprofloxacin.	Bacterial infection	[235]
ZnO-NPs (80.1–90 nm)	<i>Staphylococcus aureus</i> , <i>Salmonella Typhimurium</i> , <i>Bacillus cereus</i> and <i>Pseudomonas aeruginosa</i>	0.05 and 0.5 mg/L	ZnO NPs were used in packaging that increased safety against microbes as well as food shelf-life by inhibiting bacterial growth.	Bacterial infection	[236]
ZnO-NPs (30 nm)	<i>Campylobacter jejuni</i>	0.025, 0.03, 0.04, 0.05, and 0.10 mg/mL	Damaging membrane integrity by increasing cell membrane permeability.	Bacterial infection	[237]
ZnO-NPs (60–70 nm)	<i>S. aureus</i> and <i>P. aeruginosa</i> and standard strain of <i>E. coli</i> .	1028, 516, 256, and 125 µg/mL were used.	Producing of reactive oxygen species (ROS) is caused disruption of bacterial membranes.	Bacterial infection	[238]
ZnO-NPs (\approx 66 nm)	Eel kidney cell line (EK-1).	15.75, 31.5, and 3.15 µg/mL	Decreasing cell viability and growth rate of microorganism.	Microbial infection	[239]
ZnO-NPs (15 nm)	<i>S. pneumoniae</i>	12 µg/mL	Reducing in microbial biofilm formation.	Bacterial infection	[240]
<i>Antifungal Effects</i>					
ZnO-NPs (12–14 nm)	Aspergillus and Penicillium	5, 10, 15, 20, and 25 g/mL	Cell membrane is damaged and growth rate is inhibited by interaction of zinc ion with cell membrane.	Fungal infection	[226]
ZnO-NPs (35–129 nm)	<i>Candida parapsilosis</i>	15.65 µg/mL	Growth is inhibited and surface damage is pronounced.	Fungal infection	[236]
ZnO-NPs (70 ± 15 nm)	<i>Botrytis cinerea</i> and <i>Penicillium expansum</i>	0, 3, 6, and 2 mmol/L	Fungal hyphae is deformed, while development of conidiophores and conidia are prevented.	Fungal infection	[241]

Table 6. *Cont.*

Nanoparticles (Diameter)	Test Medium	Concentration	Effect/Result	Disease Against	References
ZnO-NPs (430 nm)	<i>Candida albicans</i> , <i>A. niger</i> and <i>A. terreus</i> .	30, 60, and 90 μL	Leading to the death of fungal hyphae by deforming of fungal hyphae.	Candidiasis, athlete's foot, mycosis, and ring worm	[242]
ZnO-NPs	<i>Candida albicans</i>	5, 10, 15, and 20 mg/mL	Producing reactive oxygen species (ROS), for example hydrogen peroxide, superoxide anion, hydroxyl radical, and hydroxyl ion.	Fungal infection	[243]
ZnO-NPs (76.15 nm)	<i>Alternaria alternata</i> , <i>Botryotis cinerea</i> , <i>Aspergillus niger</i> , <i>Penicillium expansum</i> , and <i>Fusarium oxysporum</i> .	256 $\mu\text{g}/\text{mL}$	Disruption of fungal membrane and inhibition of fungal growth.	Fungal infection	[244]
ZnO-NPs (27 ± 5 nm)	<i>Aspergillus flavus</i> and <i>Aspergillus niger</i>	25 $\mu\text{g}/\text{mL}$	Inhibiting the growth of fungus.	Fungal infection	[245]
ZnO-NPs (60 nm)	<i>Trichophyton mentagrophyte</i> , <i>Microsporum canis</i> , <i>Candida albicans</i> , and <i>Aspergillus fumigatus</i>	40 mg/mL	Inhibiting the growth of fungus.	Ring worm	[246]
ZnO-NPs (≤ 50 nm)	<i>Aspergillus fumigatus</i> Fungus and <i>Candida Albicans</i>	3, 6, and 12 mL/L	Lowering the growth rate of fungus.	Fungal infection	[247]
ZnO-NPs (13.92 nm)	<i>Alternaria alternata</i>	20–160 mg/L	The mycelia growth is inhibited.	Early blight disease	[248]
CS-Zn-CuNCS (16.6–100 nm)	<i>A. alternata</i> , <i>R. solani</i> , and <i>B. cinerea</i>	90 $\mu\text{g}/\text{mL}$	Inhibiting growth by in vitro application.	Fungal infection	[249]
<i>Antidiabetic Effects</i>					
ZnO-NPs (≤ 10 nm)	Streptozotocin-induced type 1 and 2 diabetic rats	1, 3, and 10 mg/kg	Glucose tolerance was improved, higher serum insulin (70%) and blood glucose (29%) was reduced. Nonesterified fatty acids and triglycerides was also reduced.	Diabetes	[250]
ZnO-NPs (80–100 nm)	Diabetic rats	1, 3, and 10 mg/kg	Glucose disposal, insulin levels, and zinc status are increased.	Diabetes	[251]
ZnO-NPs (10 to 20 nm)	Alpha-amylase	13.085434 $\mu\text{g}/\text{mL}$	The activity of α -amylase is inhibited.	Diabetes	[252]
ZnO-NPs (<100 nm)	Mice	8 and 14 mg/kg	Decreasing blood glucose.	Diabetes	[253]

Table 6. *Cont.*

Nanoparticles (Diameter)	Test Medium	Concentration	Effect/Result	Disease Against	References
ZnO-NPs (22.6 nm)	Wistar rats	70 mg/kg	Hyperlipidemia is controlled through lowering the levels of lipids and lipoproteins in the blood plasma.	Diabetes	[254]
ZnO-NPs	Albino rats	10 mg/kg	Ameliorative effect.	Diabetes	[255]
<i>Miscellaneous Effects</i>					
Zn-NPs (1–100 nm)	Layer chicks	30 ppm	Increasing the level of serum glucose and alkaline phosphate, while decreasing alanine transferase.	Increased chicken growth rate	[212]
ZnO-NPs (48.2 nm)	<i>Xanthomonas oryzae</i>	16.0 µg/mL	The bacterial membrane is collapsed and ruptured by interacting with ZnO NPs and as a result in the leakage of bacterial cytoplasm.	Leaf blight	[256]
ZnO-NPs (\leq 40 nm)	Rats	10 mg/kg	Heart injury is induced by ionizing radiation (IR).	Cardiovascular disorders	[257]
Zn-NPs (50–100 nm)	Swiss albino rats	10 mg/kg	Controlling blood glucose level.	Testicular diabetic complications	[258]
Vacuoles-ZnAA-NPs (AA = ascorbic acid)	B16F10 (KCLB 80080) cells. Used African-American, Asian, White donors' tissues.	ZnAA-Vac treated for 12 days at 100 and 1000 ppm	It had a stronger depigmenting impact, reducing the melanin hue by 75%.	Melanin treatment	[259]

2.5. Therapeutic Interventions of Nickel Nanoparticles (Ni-NPs)

Ni-NPs have anticancer action [260,261]. A complex structure of Qu-PEG–NiGs (48–72 nm), green synthesized by *Ocimum sanctum* leaf extract, showed mitochondrial-mediated apoptosis against the MCF-7 cell line [262], antimicrobial activity, antioxidant action, and activity against human ovarian cancer, liver and spleen injury [260,263–265], lung inflammation [266], human lung cancer [267], lymphatic filariasis [268], and larvicidal parasitic activity [269]. Bacterial protein leakage induced by ROS activation [270] and disruption of the cell membrane [271] is one way of causing bacterial cell death. The antimicrobial mechanism is shown in Figure 9. It has numerous other therapeutic properties in a single formulation or a complex formulation, as shown in Table 7.

Nanotherapeutic Application of Nickel

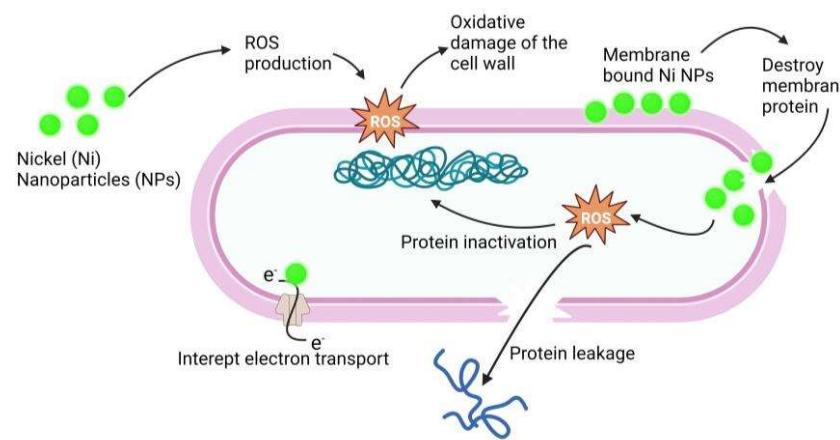


Figure 9. Antimicrobial mechanism of action of Ni-NPs. Ni-NPs cause ROS production that cause oxidative damage of the cell wall and destroy the membrane. ROS cause protein leakage and interrupt electron transport; these processes result in the antimicrobial effect of Ni-NPs.

Table 7. Nanotherapy of nickel nanoparticles.

Nanoparticles (Diameter)	Test Medium	Concentration	Effect/Result	Disease Against	References
Anticancer Effect					
DPMC-Ni-NPs (55 nm)	MCF-7, HepG2, A549, NHDF, and MTT cell lines	25, 22.47, 25.11 and 64.23 µg/mL concentration were used.	Cytotoxicity against breast cancer cell line (MDA-MB-231) was concentration dependent.	Breast cancer	[224]
Ni-NPs (1–100 nm)	Leukemia cancer cells	-	Increasing cell membrane permeability and promoting intracellular absorption in cancer cells.	Anticancer	[261]
Antibacterial Effects					
Qu-PEG-Ni-NPs (48–72 nm)	MCF-7 cells	6.25 and 50 µg/mL	Mitochondrial-mediated apoptosis is induced through ROS overproduction.	Breast cancer	[262]
Ni-NPs@E. officinalis (16.85–49.04 nm)	PA-1, SK-OV-3, Caov-3, and SW-626 cell lines were used	375, 225, 246, and 279 µg/mL	Reducing viability of malignant ovarian cell line.	Ovarian cancer	[265]
NiO-NPs (5.46 nm)	Human lung cancer cell line (A549)	93.349 µg/mL	Cytotoxicity is exhibited.	Lung cancer	[267]
Nickel-Ferrite (NiFe ₂ O ₄) nanorod, rosemary leaves used to prepare NPs, 40–200 nm	Human breast cancer (MCF-7) cell lines were used.	2, 4, 8, 16, 32, 64, 128, 256 and 512 µg/mL	NiFe ₂ O ₄ NP had cytotoxicity effect on MCF-7.	Anticancer	[272]
Penetrating the bacteria and damaging them by interacting with phosphorous- and sulphur-containing compounds such as DNA.					
Ni-NPs (30 nm)	<i>Pseudomonas aeruginosa</i> , <i>Staphylococcus aureus</i> , and <i>Klebsiella</i> sp.	2.5, 5, 10, 15 and 20 µg/mL	Penetrating the bacteria and damaging them by interacting with phosphorous- and sulphur-containing compounds such as DNA.	Bacterial infection	[263]
NMMNPs (300 to 800 nm) (NMMNPs = nickel magnetic mirror nanoparticles)	<i>S. aureus</i> and <i>E. coli</i> in <i>S. aureus</i>	0.01 g	Bacterial growth is inhibited and bacteria are killed.	Bacterial infection	[264]
NiGs-NPs (12–36 nm) Gs:green synthesized	<i>K. pneumoniae</i> , <i>E. coli</i> , <i>S. typhi</i> , <i>B. subtilis</i> , and <i>S. epidermidis</i>	25–100 µg/mL	Induced ROS generation.	Bacterial infection	[270]
Ni-NPs (0.5 nm)	<i>Staphylococcus aureus</i> , <i>Klebsiella pneumoniae</i> , <i>Pseudomonas aeruginosa</i> , <i>Vibrio cholerae</i> , and <i>Proteus vulgaris</i> .	1–0.125 mg/mL	Microbial growth inhibition.	Bacterial infection	[273]
Ni-NPs (40–80 nm)	<i>Escherichia coli</i>	21, 29 and 36 µM	Growth is inhibited.	Bacterial infection	[274]

Table 7. Cont.

Nanoparticles (Diameter)	Test Medium	Concentration	Effect/Result	Disease Against	References
Ni-NPs (10 nm and 50 nm)	Staphylococcus aurous and Escherichia coli were used.	0.42 and 0.21 µg/mL, 0.84 and 0.42 µg/mL	Destroyed bacterial cells.	Bacterial infection	[275]
Ni-NPs (<100 nm)	S. aureus and Escherichia coli	0.05, 0.1, and 1 mg/mL	Inhibited the growth of bacterial biofilm.	Bacterial infection	[276]
Miscellaneous Effects					
DPMC-Ni-NPs (55 nm)	DPPH, hydrogen peroxide, and super oxide	-	-	Oxidative stress	[260]
Ni-NPs@F. officinalis (16.85–49.04 nm)	DPPH free radicals	253, 145, and 107 µg/mL	DPPH is inhibited by adding radical species.	Oxidative stress	[265]
Ni-NPs (50 nm)	Sprague Dawley rats	1, 10, and 20 mg/kg concentrations were used.	Increasing number of WBC.	Liver and spleen injury, lung inflammation	[266]
NiO-NPs (5.46 nm)	α-amylase enzyme	268.13 µg/mL	Inhibited α-amylase enzyme and produced a hypoglycemic effect.	Diabetes	[267]
Ni-NPs (80–100 nm)	Culex quinquefasciatus.	250, 500, and 1000 ppm	Larvicidal effect.	Lymphatic filariasis	[268]
Ni-NPs (150 nm)	Larvae of R. (B.) microplus, H. a. anatolicum, C. quinquefasciatus, A. subpictus, and C. gelidus.	10.17, 10.81, 4.93, 5.56, and 4.94 mg/L	Caused larvae death.	Parasitosis	[269]
NiFe ₂ O ₄ /C nanocomposite.	In vitro: C540 (B16/F10) cells; in vivo: mice model (intratumorally injected)	1.0-MHz radiation was applied with 100 µg/mL NPs	NPs and radiation can recover tumor cells and necrosis, up to 60%.	Sondynamic therapy	[277]
NiO-NPs preparation with Neem leaf extract. (12 nm)	S. aureus and E. coli	-	Antibacterial effect was found concentration dependent.	Antibacterial	[278]
NiFe ₂ O ₄ nanoparticles (chitosan- and PEG-coated nickel ferrite), 2–58 nm	Mössbauer spectroscopy	Temperature value from 200–800 °C	Hyperthermia heating requires specific particle size, shape, magnetism, and solution concentration.	Hyperthermia heating	[279]
Nickel silicate nanoplatforms (LNS NPs)	Mouse model	-	LNS NPs may produce enough superoxide radicals when exposed to a 660 nm laser; it may simultaneously form oxygen and create superoxide radicals (O ₂ •).	Hypoxic tumor therapy	[280]

2.6. Therapeutic Interventions of Iron Nanoparticles (Fe-NPs)

Among the Fe-NPs, prominently used NPs include magnetite (Fe_3O_4), hematite, or iron (III) oxide (Fe_2O_3), and the less abundant iron (II) oxide (FeO) [281]. Magnetite (Fe_3O_4) NPs are used in biomedical applications due to their magnetic characteristics, biocompatibility, and, in particular, their superparamagnetic capabilities [282].

Magnetic NPs, also known as superparamagnetic iron oxide, are used in drug delivery [283,284] and hyperthermia therapy [285–287]. Magnetite NPs can produce reactive oxygen species (ROS), which kill microbes, making them a promising contender for an antimicrobial agent. Lung cancer cells terminated by ferroptosis as a result of Zerovalent Fe-NPs (ZVI-NPs) induce mitochondrial malfunction, intracellular oxidative stress, and lipid peroxidation; here, AMPK/mTOR activated by ZVI-NPs cause upregulation of GSK3/β-TrCP, which results in NRF2 degradation and ultimately results ferroptosis, which causes cancer cell damage [288–292], as shown in Figure 10.

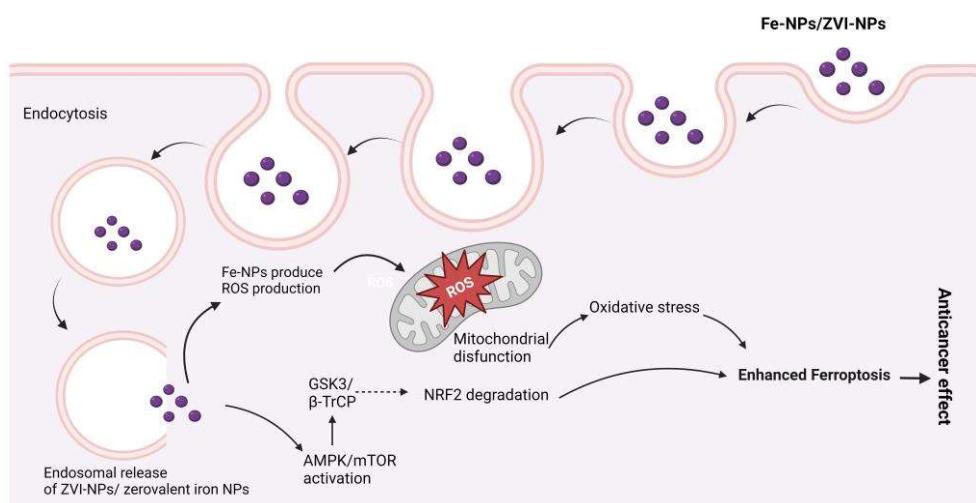


Figure 10. Possible anticancer mechanisms of iron (Fe) nanoparticles (zerovalent Fe-NPs cause ROS production, AMPK/mTOR activation, NRF2 degradation by GSK3/β-TrCP, and mitochondrial dysfunction, which results in ferroptosis).

Superparamagnetic iron oxide nanoparticles (SPIONs) provide action against the human breast cancer cell MCF7 [284]. Different nanotherapeutic studies of Fe-NPs are arranged in Table 8.

In the treatment of different types of cancer, ferroptosis, a new Fe- and ROS-dependent form of controlled cell death, has received a lot of attention. The potential of ferroptosis in combination with NPs for cancer therapy is becoming more and more clear as a result of the development of nanomaterials [293]. After cells consume Fe-based NPs, an excess of iron ions released from the lysosome in an acidic environment activates the Fenton reaction, which causes ROS formation and cell ferroptosis [294].

Importantly, when antibiotic drugs are coupled with the iron nanoparticles of neem extract, the dose of traditional antibiotics can be decreased by nearly half without affecting efficiency. As a result, the use of natural antibiotics aids in the reduction of regular antibiotic doses [295]. There was also a trial of producing bimetallic NPs (Ag-Fe) that established the synergistic antibacterial (bactericidal) impact of the two metals forming the bimetallic nanoparticles when compared to the effects of the monometallic nanoparticles against yeast and both Gram-positive and Gram-negative multidrug-resistant bacteria [296].

Nanotherapeutic Application of Iron

Table 8. Nanotherapy of Fe-NPs.

Nanoparticles (Diameter)	Test Medium	Concentration	Effect/Result	Disease Against	References
<i>Anticancer Effects</i>					
Superparamagnetic iron oxide nanoparticles (SPIONs) (7.3, 15.1, and 30.0 nm)	Human breast cancer cell MCF7	80 µg/mL	Higher measurements and more reasonable size of SPIONs upgraded the take-up sum into MCF7 cells.	Breast cancer	[162]
SPIONs	Liver cancer cells (HepG2) (in vitro)	100 µg/mL	A potent cytotoxicity on HepG2 under hyperthermia condition.	Cancer	[287]
Zero valent iron NPs (ZVI-NPs) (97.1–55.77 nm)	BALB/c mice, age: 5–6 week	5 and 10 µg/mL	Lung cancer cells died by ferroptosis as a result of ZVI-NP-induced mitochondrial malfunction, intracellular oxidative-stress, and lipid-peroxidation.	Lung cancer	[288]
CA-coated Fe ₃ O ₄ NPs (50 nm)	4T1 cells	10 µg/mL	Induced tumor cell ferroptosis.	Breast cancer	[297]
SPION-deferasirox	AS1411 DNA aptamer	100 mg	In vivo tumor growth inhibitory effect.	Antitumor	[298]
Rosemary-Fe-NPs (100 nm)	4T1 and C26 cancer cell lines	3.12 to 200 µg/mL	Rosemary-Fe-NPs exerted more cytotoxic effect.	Anticancer	[299]
CAP and iron oxide-based magnetic NPs (MNP _s)	A549 cells in vitro	50 emu/g	Potentially inhibited tumor growth.	Lung cancer	[300]
SPIONs (44.6 nm)	Breast cancer cell lines T-47D, BT-474, MCF7, and MDA-MB-231	25, 50 and 75 µg/mL	Extremely moderate molecule take-up and low cytotoxicity, while SPIONLA meaningfully affected cell take-up and cell harmfulness.	Breast cancer	[301]
Fe ₃ O ₄ @PEI-Pt(IV)-PEG-LHRH@siEZH2 nanoparticles	A2780/DDP cells (cisplatin resistant)	0.78 to 50 µM	Killing performance to A2780/DDP cells.	Anticancer	[302]
34DABA coated SPIOs (less than 20 nm)	HepG2 liver cancer cells	5, 10, 15, 20, and 25 µg	Good cytocompatibility and higher killing efficiency.	Liver cancer	[303]
Fe-NP nanopowder (35–45 nm)	PC12 cell nervous system (in vitro)	100 µg/mL	Fe-NPs induced apoptotic cytotoxicity.	Cancer	[304]

Table 8. Cont.

Nanoparticles (Diameter)	Test Medium	Concentration	Effect/Result	Disease Against	References
Au-Fe ₃ O ₄ -NPs (20.8 nm)	MCF-7 cells	50 µg/mL	Effective and promising photothermal therapy.	Breast cancer	[304]
Iron oxide nanoparticles (Fe ₂ O ₃ -NPs) (20 to 60 nm)	Lung cancer cell (A549) lines	Highest concentration of adsorbent (50 mg/L)	No toxicity against A549 cell lines.	Lung cancer	[305]
Miscellaneous Effects					
Magnetic Fe ₂ O ₃ -NPs (50–110 nm)	<i>S. aureus</i>	DMF arrangement with 40 and 60 MJ laser fluencies showed the most noteworthy antibacterial action.	ROS disrupting bacterial cell membrane.	Bacterial infection	[289]
Fe ₂ O ₃ -NPs (10–15 nm)	A/Puerto Rico/8/1934H1N1 influenza virus strain (PR8-H1N1)	1.1 pg	Inactivation of cell protein through the communication of nanoparticles and -SH bunch (proposed, not examined at this point).	Viral infection	[291]
Fe ₂ O ₃ -NPs (10–15 nm)	H1N1 Influenza A	4.25 ± 0.2 pg	Change in viral RNA transcripts within 24 h, eight-fold reduction when treated with iron oxide.	Viral infection	[291]
Fe ₂ O ₃ -NPs (10–30 nm)	<i>Trichothecium roseum</i> , <i>Cladosporium herbarium</i> , <i>Penicillium chrysogenum</i> , <i>Alternaria alternate</i> , and <i>Aspergillus niger</i>	0.063–0.016 mg/mL	Development of ROS, protein and DNA damage oxidative stress was the way of producing antifungal effect.	Fungal infection	[292]
Zero-valent iron (Fe0) NPs, spherical (31.1 nm)	<i>Staphylococcus aureus</i> (Gram-positive) and <i>E. coli</i> (Gram-negative)	MIC at 30 µg/mL and complete growth inhibition concentration at 60 µg/mL	Oxidative stress generation via ROS and visible damage to bacterial protein and DNA.	Bacterial infection	[298]
4 nm core Fe ₂ O ₃ coated with tartric/adipic acid	Mitochondrial DNA (mtDNA), mitochondrial function, and autophagy in colorectal cell lines (HT-29)	0.5 mM/L	Reduced the number of mtDNA copies (indicative of a reduction in the number of mitochondria in these tumor cells).	Mitochondrial dysfunction	[306]

3. Metal Nanoparticles Elimination from Body

The elimination of NPs depends on their particle size, intrinsic biodegradability, core density, surface charge, and surface chemistry [307]. The liver is the major clearance organ in the oral administration of NPs. Intravenously administered NPs are cleared from the bloodstream by two main mechanisms: (i) renal elimination and (ii) hepatobiliary elimination. Choi et al. [308] reported that smaller-sized (<5.5 nm diameter) quantum dots undergo efficient urinary excretion due to the pore size limit of glomerular filtration in the kidneys. According to estimates of Si-NPs in rats, 7–8% of NPs were eliminated in urine and 75–80% were expelled in feces [309]. Nonbiodegradable and larger-sized (>5.5 nm) NPs are supposed to be eliminated through the hepatobiliary route. The hepatobiliary elimination involved the following pathways: (1) the liver sinusoid; (2) the space of Disse, a tiny perisinusoidal space containing blood plasma, nutrients, oxygen, and body waste that has become crucial in the treatment of liver disease, which is located between endothelial cells and hepatocytes; (3) hepatocytes; (4) bile ducts; (5) intestines; and finally (6) out of the body, as shown in Figure 11. In hepatobiliary elimination, the liver nonparenchymal cells (e.g., Kupffer cells and liver sinusoidal endothelial cells) influence and determine the elimination fate. The removal of Kupffer cells increased the fecal elimination of NPs by more than 10-fold [310].

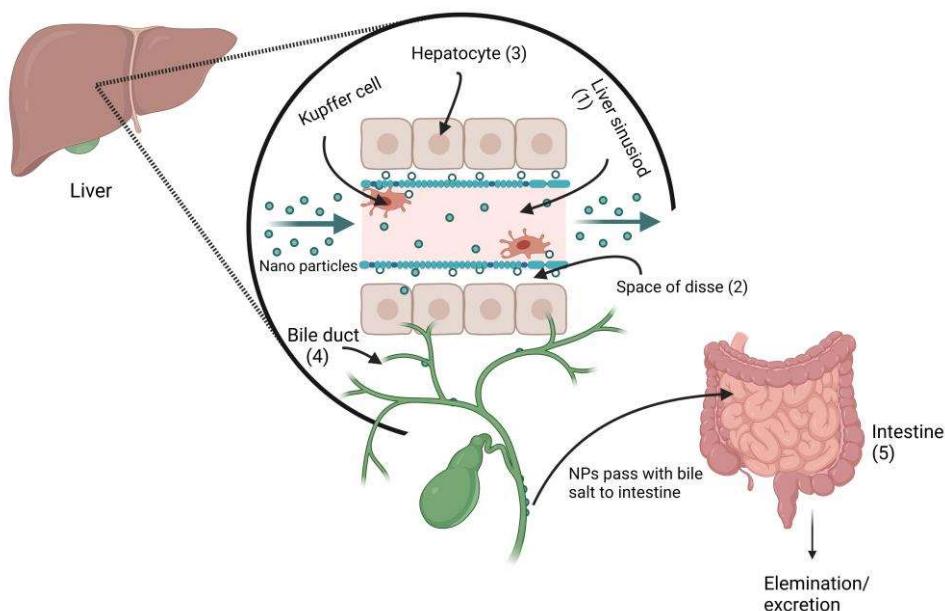


Figure 11. Proposed metal nanoparticles hepatobiliary clearance pathway (when metal NPs pass through the liver sinusoid, they enter the space of Disse via Kupffer cells, and then enter the bile duct, followed by fecal elimination).

NPs can enter the body through multiple routes, including the skin, respiratory tract, dermal exposure, mucosal, oral, intravenous, subcutaneous, intramuscular, etc., and can induce acute or chronic toxicities [311]. The anionic NPs are less toxic than the cationic NPs, which cause hemolysis and clotting [312]. Singh et al. [313] reported that ceramic NPs, commonly used for drug delivery, exhibit oxidative stress and cytotoxic activity in the lungs, liver, heart, and brain, as well as having teratogenic or carcinogenic effects. NPs have been shown, both *in vivo* and *in vitro*, to increase cellular reactive oxygen species, induce multiple minor and severe toxicities, and even disrupt host homeostasis [311]. Although NPs are useful for numerous medical applications, there are still some concerns for ecosystems and living organisms due to their uncontrollable use and discharge to the natural environment; thus, it should be considered to make the use of NPs more convenient and environmentally friendly. Preclinical studies have revealed the importance of renal-clearable luminous metal NPs in cancer therapy, which offers tremendous promise for

potential clinical translation [314]. The retention of NPs in the body, especially in the vital organs, usually depends on the density of the particles. In a study of gold and silver NPs by Tang et al., it was demonstrated that the lower-density metal NPs have a higher distribution and shorter retention time than the higher-density metal NPs [315].

4. Conclusions

Nanodrugs can be highlighted as the future of medicine, and using potential metals such as Fe, Au, Cu, Ag, Ni, and Zn in NPs showed optimistic results against various types of cancers, as well as displaying antitumor, antidiabetic, and antimicrobial activity. They are also applicable for other purposes, and it was found that metal NPs have significant synergistic activity with commercially available antibiotics. Since we already have certain levels of most metals in our bodies, they are compatible with our immune systems, which is of benefit to metal nanotherapy. Research has found that metals enhance the pharmacological activity considerably.

Despite recent advances in metal nanotherapy, the majority of nanotherapeutics are still being studied. The main concerns are not only their long-term safety for the patient, but also the ecological and toxicological aspects that need to be considered.

The generation of ROS is a significant challenge for metal NPs and metal oxide NPs. Diameter, structure, interface, content, solubility, accumulation, and particle absorption are factors that can affect ROS generation. A metallic nanomaterial's toxicity may vary based on its oxidation reaction, ligand, solubility, shape, environment, and medical factors. For example, characterization and cell type are important factors in the uptake of Au-NPs. If the Au-NPs are absorbed by a healthy cell, they will eventually be removed, but if they are absorbed by a malignant cell, they will cause cell death. More *in vivo* metal nanotherapeutic studies are needed to find out the toxicological conditions in normal cell lines when targeting cancer cells.

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